



**Everything You Ever Wanted  
to Know About CT  
(and then some)**

**Rob Pine & Joe Savage  
NMED/DWB**

# Background

- All surface water contains bacteria and may contain protozoa and viruses.
- Many waterborne pathogens, i.e. organisms harmful to human health, are a result of contamination of surface water bodies.
- Pathogens can also contaminate a groundwater supply or enter into a distribution system.
- Prior to water disinfection and filtration, waterborne pathogens killed thousands/yr in the U.S.

# Background

Waterborne outbreaks in U.S. public water systems between 1991 and 2000

<b>Pathogen</b>	<b># Outbreaks</b>	<b># Cases</b>
E. coli O157:H7/C. jejuni	1	781
Giardia	16	2,240
Hepatitis A virus	1	46
Norwalk-like viruses	3	2,400
Cryptosporidium	9	408,220
Salmonellae (nontyphoid)	2	749

# Reg Requirements

- SWTR: all surface water & GWUDI systems required to achieve 99.9% (3-log) removal and/or inactivation of Giardia and 99.99% (4-log) inactivation of viruses through treatment and disinfection.
- Groundwater Rule: Groundwater systems with a source water sample that is fecal-positive or with other significant deficiencies may be required to disinfect or otherwise treat to achieve 99.99% (4-log) inactivation of viruses.



# Disinfection

Different disinfectants have advantages and disadvantages. Chlorine ( $\text{Cl}_2$ , calcium hypochlorite or sodium hypochlorite) are most common in NM.

<b>Disinfection</b>	<b>Advantages</b>	<b>Disadvantages</b>
Chlorine	<ul style="list-style-type: none"><li>• Leaves residual</li><li>• Effective w/bacteria, viruses and Giardia</li></ul>	<ul style="list-style-type: none"><li>• Creates DBPs</li><li>• Not effective against Crypto</li></ul>
Chloramines	<ul style="list-style-type: none"><li>• Lower DBP production</li></ul>	<ul style="list-style-type: none"><li>• Relatively weak disinfectant</li><li>• Requires using Ammonia</li></ul>
Chlorine dioxide	<ul style="list-style-type: none"><li>• Effective against Crypto</li><li>• Will not form THMs or HAAs</li></ul>	<ul style="list-style-type: none"><li>• Highly volatile residuals</li><li>• High chemical cost</li></ul>
UV	<ul style="list-style-type: none"><li>• No chemicals</li><li>• Effective against Crypto</li><li>• No DBPs</li></ul>	<ul style="list-style-type: none"><li>• Doesn't leave residual</li><li>• High energy consumption</li></ul>

# Chlorine Disinfection

Chlorine does not kill pathogens instantaneously on contact. The rate of inactivation depends on 5 factors:

- The Pathogen
- Chlorine Concentration,  $C$  (mg/l)
- Contact Time,  $T$  (minutes)
- Temperature of water
- pH of water

# Pathogens

Some pathogens are harder to inactivate with chlorine than others

- Bacteria are easily inactivated
- Viruses are easily inactivated
- Giardia is tougher to inactivate and requires more contact time or higher residual Chlorine
- Chlorine is not very effective at inactivating Cryptosporidium

# Chlorine Concentration & Time

- For bacteria, viruses and Giardia, the inactivation rate increases with contact time,  $T$ , or with increasing chlorine concentration,  $C$
- In order to determine if  $C$  and  $T$  are adequate to get a certain % inactivation for a given pathogen, one can consider  $C \times T$ , usually written  $CT$ , and compare the measured value with minimum requirements for  $CT$



# Calculating CT

- Use conservative estimates:
  - For T, use time estimated between disinfectant injection and first customer based on peak flow rates
  - For C, use lowest measured daily residual at injection point
- T is measured in minutes, C is measured in mg/l, Temperature is measured in °Celsius
- If you have adequate CT for Giardia, then you have adequate CT for viruses.

# Calculating CT

To Calculate T, you must find the volume of each section of distribution and storage between the disinfection injection points and the first customer:

- For storage, use the lowest water level
- volume of a cylinder (pipe, tank):  $v = H \times \pi \times r^2$   
where H = pipe length or minimum water level  
r = radius
- Convert cubic feet (ft<sup>3</sup>) into gallons by multiplying by 7.48 gallons/ ft<sup>3</sup>
- For a given unit,  $T = v / (\text{maximum flow rate})$

# CT Example 1

**Problem:** Groundwater system injects chlorine 1000 feet before the first customer. No storage tank. Distribution line is 6 inch diameter. Maximum pumping rate is 150 gpm. Lowest chlorine residual (C) measured is 1.2 mg/l. Temperature of the water is 11 °C and the pH was 7.1

## Solution:

1. Cross-sectional area of pipe:  
 $\pi (.25 \text{ ft})^2 = .196 \text{ ft}^2$
2. Volume of length of pipe:  
 $1000 \text{ ft} \times .196 \text{ ft}^2 = 196 \text{ ft}^3$
3. Convert pipe volume to gallons:  
 $196 \text{ ft}^3 \times 7.48 \text{ gal/ ft}^3 = 1,469 \text{ gal}$



# CT Example 1

Solution (continued):

4. Calculate time between injection point and first customer:

$$T = 1,469 \text{ gal} / 150 \text{ gallons per minute} = 9.79 \text{ minutes}$$

5.  $CT = 1.2 \text{ mg/l} \times 9.79 \text{ min} = 11.7 \text{ mg-min/l}$

Note: Units of CT are meaningless





# Bonus Question

What pathogen is very resistant to inactivation by Chlorine?

# CT Tables

How do we know if the calculated CT value is good enough?

- EPA has published CT tables for different pathogens and disinfectants
- The tabulated values in the CT tables are also dependent on temperature and pH.

# CT Tables

Below is a CT table for viruses. Note how temperature and pH impacts CT:

	Log Inactivation					
	2.0		3.0		4.0	
pH	6 - 9	10	6 - 9	10	6 - 9	10
Temperature (°C)						
0.5	6	45	9	66	12	90
5	4	30	6	44	8	60
10	3	22	4	33	6	45
15	2	15	3	22	4	30
20	1	11	2	16	3	22
25	1	7	1	11	2	15

# Rules of Thumb

Note from CT tables that:

- Chlorine is more effective with decreasing pH
- Chlorine is more effective with increasing Temperature
  - May be easier to meet CT in summer than winter
  - DBP production increases with higher temp
- Use most conservative value from table for temperature and pH values in between tabulated values



# Example 1 (continued)

**Problem:** Does the system in Example 1 meet the 4-log reduction requirement for Viruses through chlorine disinfection?

**Solution:**

6. Find required CT:

Using the CT table for viruses, at temp = 11 °C, pH = 7.1 and % reduction = 99.99% (4 log), we find the CT value in the table is 6

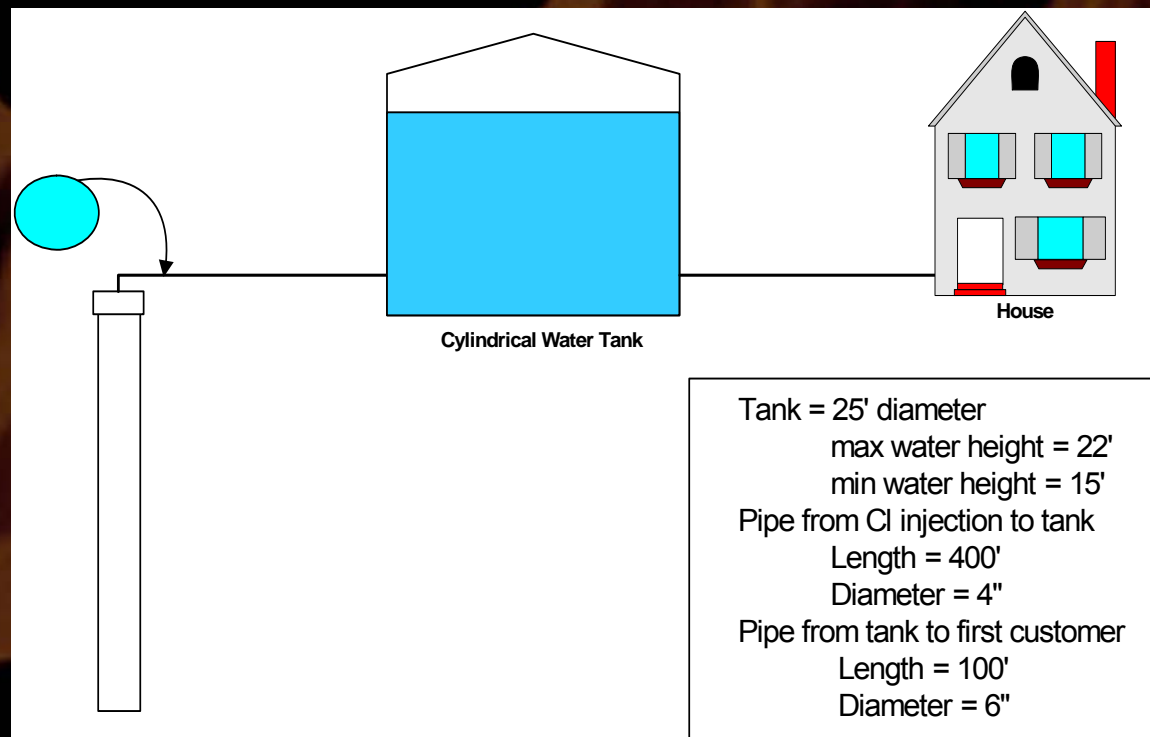
7. The value calculated for the system was 11.7. Because  $11.7 > 6$ , the CT at the system was adequate to meet the CT requirement of 4-log reduction of viruses.

# Residual Monitoring

- Surface Water systems that filter must monitor chlorine residual of water entering distribution continuously and the lowest daily value recorded. Daily pH and temp must also be recorded.
- Groundwater systems required to disinfect under the GWR must monitor residual as follows:
  - If system serves  $> 3,300$  customers, monitor continuously and record lowest daily value.
  - If system serves  $\leq 3,300$  customers, system takes daily grab sample at time of peak flow.
  - No explicit requirement for pH and temp, but this is necessary to demonstrate 4-log removal.

# Example 2

**Problem:** System shown below has peak flow rate of 250 gpm. Low residual measured at .8 mg/l. pH measured at 7.5, water temp measured at 5 °C. What is CT?





## Example 2

### Solution:

1. Find T for pipe 1:

$$\text{volume} = 400\text{ft} \times \pi \times (.167 \text{ ft})^2 \times 7.48 \text{ gal/ft}^3 = 261 \text{ gal}$$

$$T = 261 \text{ gal}/250\text{gpm} = 1.04 \text{ minutes}$$

2. Find T for pipe 2:

$$\text{volume} = 100\text{ft} \times \pi \times (.25 \text{ ft})^2 \times 7.48 \text{ gal/ft}^3 = 146 \text{ gal}$$

$$T = 146 \text{ gal}/250\text{gpm} = .584 \text{ minutes}$$

Question : Is 250 gpm the correct flow for pipe 2?



## Example 2

3. What about the tank?

If you assume no short-circuiting (plug flow):

$$\text{volume} = 15\text{ft} \times \pi \times (12.5\text{ft})^2 \times 7.48 \text{ gal/ft}^3 = 55,076 \text{ gal}$$

$$T = 55,076 \text{ gal}/250 \text{ gpm} = 220 \text{ minutes}$$

Generally assume there is short circuiting in which case you use a short-circuiting factor,  $F_{sc}$ , as follows:

$$T = F_{sc} \times (\text{minimum volume})/(\text{maximum flow})$$

# Example 2

The following is a table of  $F_{sc}$  values from EPA:

<b>Baffling Condition</b>	<b>Baffling Description</b>	<b><math>F_{sc}</math></b>
Unbaffled	None; high inlet and outlet flow velocities, variable water level	0.1
Poor	Single or multiple unbaffled inlets and outlets, no intra-basin baffles	0.3
Average	Baffled inlet or outlet with some intra-basin baffling	0.5
Superior	Perforated inlet baffle, serpentine or perforated intra-basin baffles, outlet weir or perforated weir	0.7
Perfect (plug flow)	Very high length to width ratio (pipeline flow), perforated inlet, outlet, and intra-basin baffles.	1.0

## Example 2

Assuming this tank was unbaffled, we get:

$$T = 0.1 \times 55,076 \text{ gal}/250 \text{ gpm} = 22 \text{ minutes}$$

4. Total CT (with unbaffled tank):

$$\begin{aligned} \text{CT} &= .8 \text{ mg/l} \times (22 \text{ min} + 1.04 \text{ min} + .584 \text{ min}) \\ &= 18.9 \text{ mg-min/l} \end{aligned}$$

5. Do we have 4-log inactivation for viruses?

Required CT from table is 8 mg-min/l

# Surface Water CT

- Surface water systems need to have 3-log inactivation/removal of Giardia.
- Systems that pre-chlorinate can count the time through treatment system up to the point of post-chlorine injection.
- EPA has published CT tables for Giardia. These tables are much more detailed with pH in increments of .5 and CT values depending on the free chlorine residual.



# More Rules of Thumb

- Time and volume estimates must be conservative
- Floating tanks get no contact time credit

# Bonus Questions

Chlorine is more effective at inactivating Giardia with increasing \_\_\_\_\_

\*Describe groundwater system requirements for reduction of Cryptosporidium, viruses and Giardia

# Example 3

**Problem:** peak flow rate of 250 gpm. Low residual measured at .8 mg/l. pH measured at 7.5, water temp measured at 5 °C. What is CT?

**Solution:** Calculated CT is the same as in Example 2. However, we must now use CT tables for Giardia. The following is an excerpt:

Chlorine (mg/l)	pH = 7.5					
	Log Inactivation					
	0.5	1.0	1.5	2.0	2.5	3.0
<= .4	28	55	83	111	138	166
0.6	29	57	86	114	143	171
0.8	29	58	88	117	146	175
1	30	60	90	119	149	179
1.2	31	61	92	122	153	183
1.4	32	62	94	125	156	187

## Example 3

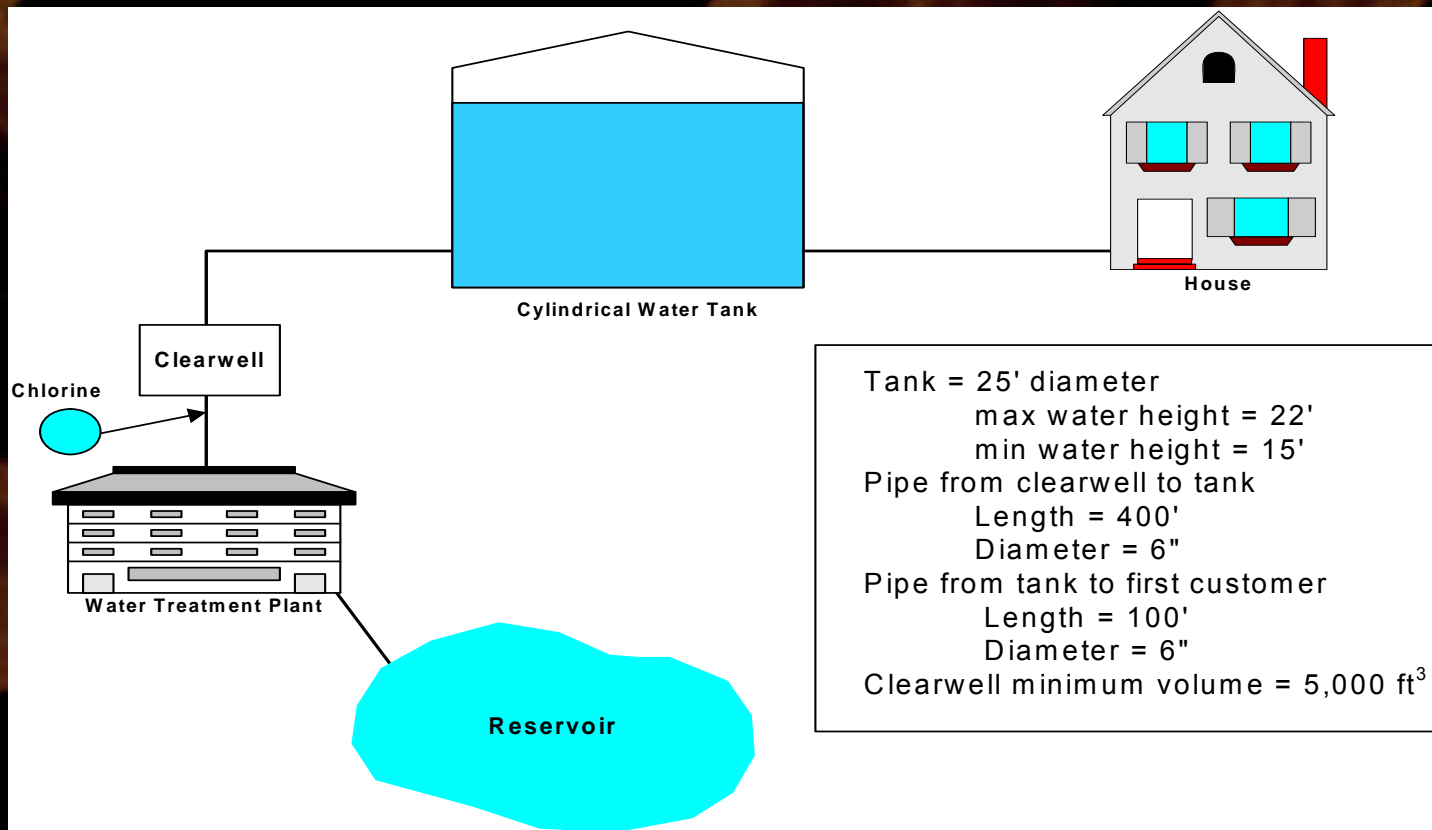
The system doesn't meet required CT for giardia through disinfection alone. What are some things the system could do to meet CT requirement?

- Removal credits through treatment
- Increase disinfectant residual
- Add baffling to tank
- Increase minimum water level in tank
- Add storage capacity



# Example 4

**Problem:** System shown below has peak flow rate of 250 gpm. Low residual measured at .8 mg/l, pH measured at 7.5, water temp measured at 5 °C. What is CT? If the system gets 2 log removal credit from treatment, does it have adequate CT?



# For More Information

- Contact Rob Pine      505-476-8642  
[robert.pine@state.nm.us](mailto:robert.pine@state.nm.us)
- Contact Joe Savage      575-258-3272  
[joe.savage@state.nm.us](mailto:joe.savage@state.nm.us)
- Look at EPA's LT1ESWTR Disinfection Profiling and Benchmarking guidance document at  
<http://www.epa.gov/safewater/mdbp/pdf/profile/lt1profiling.pdf>